2017-10

As per the New Semester-wise Syllabus of Gondwana University

BOTANY

B.Sc. Semester IV (CBCS)

Paper – I Cell Biology, Genetics and Biotechnology

Paper - II Plant Ecology

Dr. Aparna S. Margonwar Dr. Sharadkumar Patit Mr. Vijay S. Khonde Miss Nilima Rangari



CONTENTS

B.Sc. SEMESTER – IV

Paper - I - Cell Biology, Genetics and Biotechnology

UNIT - I:

- Ultrastructure and function of typical plant cell: Cell wall, Plasma Membrane; General structure of Nucleus, Mitochondria, Plastids, Endoplasmic Reticulum, Golgi Complex, Vacuole, Lysosome, 1.
- Cell Division: Mitosis, Meiosis with respect to plant cells.
- DNA:Structure and replication of DNA.
- Plant Tissue culture: Concept of totipotency, Steps of plant tissue 3. culture from explant to whole plant regeneration.

UNIT - II:

- Mendelism:Laws of inheritance (Dominance, Segregation and Independent Assortment), back cross and test cross. 1.
- Interaction of genes: with reference to plants. a) Allelic interaction -Incomplete Dominance (1:2:1) b) Non-allelic interaction Complementary genes (9:7), Supplementary genes (9:3:4). . 2.
 - Extra nuclear genome: Structure and functions of Mitochondrial and Plastid DNA.

UNIT - III:

- Linkage: Definition, Gene theory of Morgan, types of linkage-Complete and Incomplete, significance.
- Crossing over: Definition, theories (Breakage and Reunion, 2.
- Variation in Chromosome number: Polyploidy (Auto- and Allo-), Aneuploidy (Nullisomy, Monosomy, Trisomy and Tetrasomy), Significance. 3.
- Structural changes in chromosome: Deletion and Deficiency, Duplication,
- Mutation: Definition, Types-Spontaneous and Induced; Substitution and Frame-shift, Mutagens-Physical and Chemical, application of Induced 5. Mutation in Crop Improvement.

UNIT - IV

- 1. Genetic Engineering: Tools and techniques of Recombinant DNA technology (RDT)- a) Cloning vectors (Plasmids - PBR 322, Bacteriophages-T4 phage, lambda Phage and Agrobacterium) b) Restriction enzymes and Ligases c) Genomic and complementary DNA (c-DNA) libraries
- 2. Protein synthesis-transcription and translation
- 3. Jumping genes (Transposons): Ac/Ds elements in Maize.
- 4. Regulation of gene action in Prokaryotes: Lac-Operon concept.

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UNIT I

ULTRASTRUCTURE AND FUNCTION OF TYPICAL PLANT CELL

Introduction

All organisms are composed of structural and functional units of life called 'cells'. The body of some organisms like bacteria, protozoans and some algae is made up of a single cell while the body of plants and animals are composed of many cells.

Anton van Leewenhock invented the microscope, Robert Hooke in 1665 observed a piece of cork under the microscope and found it to be made of small compartments which he called "cells" (Latin cell = small room). In 1672, Leewenhock observed bacteria, sperm and red blood corpuscles. In 1831, Robert Brown, an Englishman observed that all cells had a centrally positioned body which he termed the nucleus. In 1838 M.J. Schleiden and Theodore Schwann formulated the "cell theory." The cell theory maintains that all organisms are composed of cells. Cell is the structural and functional unit of life. Which arise from pre-existing cells.

A cell may have a unit of protoplasm bounded by a plasma or cell membrane and possessing a nucleus. Protoplasm is the living substance and includes the cytoplasm and the nucleus. The cytoplasm has in it organelles such as ribosomes, mitochondria, golgi bodies (in plant dictyosome), cytoplasm has in plant cell) lysosomes and endoplasmic reticulum. Plant cells have in their cytoplasm large vacuoles containing non-living inclusions like crystals, pigments etc.

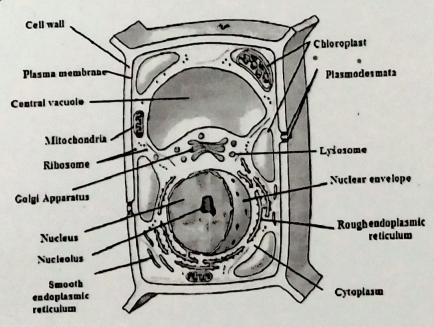


Fig. 1.1: Typical Plant Cell

As per the New Semester-wise Syllabus of Gondwana University

BOTANY

B.Sc. Semester V (CBCS)

Discipline Specific Elective-I (DSE-I)

Paper - I Genetics and Plant Breeding - I

Paper - II Genetics and Plant Breeding - II



Dr. Aparna S. Margonwar Dr. Sharadkumar P. Patil Ms. Nilima U. Rangari Dr. Sivaprasad Hari Mr. D. N. Watakhere

Himalaya Publishing House

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>	Non-Mendelian Inheritance: Lethal gene (2:1); Co-dominance (1:2:1), Dominant epistasis (12:3:1), Recessive epistasis (13:3); Inhibitory gene (15:1); Polymeric gene (9:6:1).	1 - 25
>	Multiple allelism and Pleotropism.	
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>	Cytoplasmic Inheritance: Leaf variegation in Mirabilis jalapa, Mutations in mitochondrial DNA cause human disorders, Kappa particles in Paramecium, maternal effect.	
>	Chromosome theory of inheritance.	26 - 47
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•	Sex determination and sex-linked inheritance: Sex determination in Drosophila, humans and plants, Klinefelter and Turner's syndrome, Barr bodies, Lyon's hypothesis.	
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>	Plant genetic resources, acclimatization	
>	Plant Introduction: Procedure of plant introduction -quarantine- cataloguing- evaluation - multiplication, distribution - acclimatization, purpose of plant introduction, achievements, merits and demerits.	
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>	Hybridization: history, objectives, types, procedure (Emasculation methods, bagging, tagging, pollination, harvesting and storing of F1 seeds, selfing), techniques and consequences, advantages and limitations.	

A Paris

Paper - I

Unit - I

Brief History of Mendel

Gregor Johann Mendel was born Johann Mendel on July 22, 1822, to Anton and Rosine Mendel on his family's farm in Heinzendorf, Austria. He spent his early youth in that rural background until age 11. A local schoolmaster was impressed with Mendel attitude for learning and that's why he recommended for secondary school in Troppau to continue his education. Mendel had financial problem but he excelled in his studies and in 1840, he graduated from the school with honors.

After his graduation, Mendel enrolled in a two-year program at the Philosophical Institute of the University of Olmutz. There he again distinguished himself academically particularly in the subjects of physics and math. He graduated from the program in 1843.

That same year Mendel father expected him to take over the family farm but Mendel want studying to be a monk. He joined the Augustinian order at the St. Thomas Monastery in Brno and was given the name Gregor. At that time, Mendel was immediately exposed to the research and teaching of its members. He also gained access to the monastery's extensive library and experimental facilities.



Gregor Johann Mendel (1822-1884)

In 1849, when his work in the community in Brno exhausted him to the point of illness, Mendel was sent to fill a temporary teaching position in Znaim but he failed a teaching-certification exam. Then he was sent to the University of Vienna at the monastery's expense to continue his studies in the sciences. While there, Mendel studied Mathematics and Physics under Christian Doppler, after whom the Doppler effect of wave frequency is named. He studied Botany under Franz Unger who begun using a microscope in his studies, and who was a proponent of a pre-Darwinian version of evolutionary theory.

In 1853, upon completing his studies at the University of Vienna Mendel returned to the monastery in Brno and was given a teaching position at a secondary school where he would stay for more than a decade.

Between 1856 and 1863 Mendel cultivated and tested 29,000 pea plants and carefully analyzing seven pairs of contrasting character of Pea plants. Mendel worked on this for several years carefully self-pollinating and wrapping each individual plant to prevent accidental pollination by insects. He collected the seeds produced by the plants and studied the offspring of these seeds observing that some plants breed true and others not. Mendel discovered that by crossing tall and short parent plants he got hybrid offspring that resembled the tall parent rather than being a medium height blend. He explained the concept of heredity units, now called genes. These often expressed dominant or recessive characteristics. He then worked out the pattern of inheritance of various traits and produced two generalizations that became known as the laws of heredity.



According to the new syllabus of Gondwans University, Gadchiroli.



A HAND BOOK

PRACTICAL CHEMISTRY

B.Sc. - II SEMESTER - III & IV

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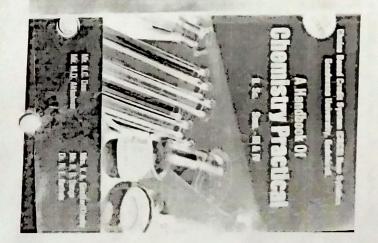
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Principal
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- 3. INORGANIC CHEMISTRY SECTION (Semester-IV) 43 to 74
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SEMESTER-III

INORGANIC CHEMISTRY

Experiment Number-1

Alm: To determine two acidic and two basic radicals in the given mixture.

Apparatus: Test tubes, pair of tongues, test tube holder, burner, wash bottle etc.

Principle: Qualitative analysis of any inorganic salt involves the identification of cation and anion in the salt.

If a mixture of two salts is given for analysis it contains four ions, two cations and two anions. In general, they are called as radicals. Positive ions are called as basic radicals while negative ions are called as acidic radicals. Hence one has to identify two acidic and two basic radicals in the mixture of two salts.

Identification of acidic radicals is done by individual tests. However, analysis of basic farlicals requires systematic procedure so that they can be identified correctly without interference of their ions.

Separation of basic radicals is based on the concepts of solubility product and common sin effect. When the ionic product of any salt exceeds the solubility product, it gets precipitated.

Hence, .it is possible to separate various ions in the form of their salt like chlorides, surphides, hydroxides, etc, by exceeding their solubility products.

flutection of Acidic Radicals (Anions)

Tests for Common Acidic Radicals

Group-I

- 1) Test for Carbonate (CO3--)
- a) Take a pinch of mixture in dry test tube and + dil. HCl or if not dil. H₂SO₄. If colorless with brisk effervescence is produced then Carbonate (CO₃⁻⁻) present.

AS PER CICS PATTERN SYLLABOS OF GONDWANA UNIVERSITY, GADCHIROLL 4099

Text Book

Inorganic Chemistry

B.Sc. Part - III (Semester - VI) Paper

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Bhagwantrao Arts & Science College, Etapalli Dist. Gadchiroli

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Unit - I

A. Qualitative & Quantita Aspects of Analysis

Definition

The process by which researchers select a representative straight population that could be studied for their topic so the treat conclusions about the entire population.

CR

Sampling is the process to get a representative and homes

Representative means that content of analytical samples

Homogeneous means that the analytical sample has the wa

- A sample is the representative of the whole bulk it reflect the composition of the bulk.
- Critical step in analysis as the significance and accents.
- Sample can be solid, liquid, gas and heterogeneous simple.

Homogenous sample: A grab sample is often ok. For interest sample (blood, urino) can be analyzed directly as it is homogen

Heterogeneous sample: several individual samples are to gepresent content of shipment of grains. One has to collect little page loading / unloading using a sampling spear (sack sample assignoss sample. N s. Rajni Prakashan & Books Distributor



As Per U.G.C. Semester Pattern New Syllabus of Gondwana University, Gadchiroli

Text Book of Chemistry

PHYSICAL CHEMISTRY



Paper-II

B.Sc. Semester-VI (CBCS)



(3398

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1 PHOTOCHEMISTRY

1.1 Introduction

000

1) "Photochemistry deals with the reaction involving absorption or en radiation. The reacting molecules absorb photons of light and get excite excited molecules ultimately yield product. Such a chemical reaction which i by absorption of ultra violet/visible radiations is called as Photochemical R

Example: $H_2 + Cl_2 \xrightarrow{hv} 2HCI$

- 2) It differs from the ordinary chemical reaction. Ordinary reaction occurs by a of heat energy and in absence of light. Such a reaction is termed as Dark rethermal reaction. eg. $N_2 + 3H_2 \rightarrow 3NH_3 \cdot H_2 + 1 \rightarrow 2HI$.
- 3) Now-a-days photochemistry has become one of the important bra research. Many chemical synthesis like vitamin D, formation of ozone, nur molecules, insecticides can be carried out photochemically.
- Fluorescence and phosphorescence phenomenon have emerging apple fluorescent tube light, TV and X-ray screens, luminescent watch dials, optical in white' cloth etc.

1.2 Interaction of Radiation with Matter

When light falls on any system, the part of incident light energy is reflected it may get absorbed and remaining may be transmitted. Photochemistry, however, the part of light which is absorbed by the matter.

AS PER CBCS PATTERN SYLLABUS OF GONDWA





Physical Chemistry

B.Sc. Part - III (Semester - V) Paper - II

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Electrochemistry - I

1.1 Introduction:

The branch of chemistry which deals with the study of relationship between electrical energy, chemical energy and interconversion of one form of electrical energy into another is called electrochemistry. In the present unit, we shall focus our attention on the study of various aspects of conductance, and its measurements.

1.2 Flectrical Conductance

All substances do not conduct electrical current. The substances which allow the passage of electric current are called conductors. The best conductors are metals such an copper, silver, tin, etc. On the other hand, the substances which do not allow the passage of electric current through them are called non-conductors or insulators. Some common examples of insulators are rubber, wood, wax, etc.

1.3 Types of Conductors

Conductors are mainly divided into two types:

- (a) Metallic conductors: These are the metallic substances which allow the passage of electricity through them without undergoing any chemical change. Some common examples are copper, silver, aluminium, etc. The conduction through metals is due to the movement of electrons in the metallic crystals.
- (b) Electrolytic conductors or electrolytes: These are the substances which allow the passage of electricity through their molten state or through their aqueous solutions and also underen themical decomposition at the same time. Some common examples of

A Boats Districtions 252

As per New Semester Pattern Syllabus
Gondwana University, Gadchiroli
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Pattern



ORGANIC CHEMISTRY

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Spectroscopy

Introduction:-

Spectroscopy is the branch of science that deals with the interaction of electromagnetic radiations with the matter Spectroscopic methods are generally used to measure the energy difference between various molecular energy levels and to determine the atomic and molecular structures. The instruments used in such studies, called spectrophotometers. These are the deviced to measure the relative energy that is emitted, transmitted or reflected in the infrared, visible or ultraviolet regions, as a function or wavelength or wavernumber.

Nuclear Magnetic Resonance (NMR) Spectroscopy

Nuclear magnetic resonance (NMR) spectroscopy is an absorption spectroscopy in which sample absorb electromagnetic radiation in the radio-frequency region (3MHz to 30,000MHz). As the name itself indicates, NMR spectroscopy involves nuclear magnetic resonance which depends on the magnetic property of atomic nuclei. Thus, NMR spectroscopy deals with nuclear magnetic transitions between magnetic energy levels of the nuclei in molecules.

NMR signals were first observed in 1945 independently by Purcell and Black. Ethanol was the first compound studied in NMR spectroscopy. In 1952, Purcell and Bloch won the nobel prize in physics for their discovery. NMR Spectroscopy is more useful for structure elucidation of organic compound than UV - visible and IR - Spectroscopy.

There are about 100 isotopes for which NMR spectroscopy is possible, but the most commonly used are proton nuclear magnetic resonance (PMR or 'H NMR) Systematopy and carbon 13 ("C NMR) Spectroscopy

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अण्णाभाक्षसाहै श्रमताधिष्ठित समाज खंगरीचे तत्वज्ञ

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साठे यांच्या ाऊ साठे यांच्या विषमता अण्णाभाऊंच्या

त्य संस्कृती

् पाटील डाळे

व कुंभार, डॉ.

ही.एच.हनवते

रायण कांबळे.

- २०१९"

''अण्णाभाऊ साठे : दलित साहित्य आणि आंबेडकरी चळवळ''

तुळशिराम शंकर कांबळे संशोधक, एम.ए.मराठी, अर्थ, बि,एड, बी.पी.एड मु. मोदुमतुर्रा पो. इंदाराम ता. अहेरी जि. गडचिरोली. प्रा. सुधीर टि. भगत, मार्गदर्शक, एम.ए.मराठी, पि.एच.डी, (मराठी विभाग प्रमुख) भगवंतराव कला/विज्ञान महाविद्यालय एटापल्ली

अण्णा भाऊ साठे एक दिलत साहित्यिक असून दिलत साहित्य प्रवाह हा परिवर्तनाच्या चळवळीतुन उदयास आलेला आहे. जीवनाची बांधिलकी मानणारे दिलत साहित्य हे जीवनातले वास्तव प्रश्नांना घेऊन समोरा जात आहे. दिलतांच्या आयुष्यातील विविध वचना, वेदना, दुःख, दैन्य, दास्य, आणि जीवनानुभव हेच साहित्याच्या केंद्रस्थानी आहे. आज बाबासाहेबांच्या अनुयायांनी ''दिलत'' हा शब्द नाकारून प्रस्तुत साहित्याला आंबेडकरी साहित्य आणि दिलत चळवळीला आंबेडकरी चळवळ असे नामाभिधान दिले आहे. हा शब्द प्रयोग निःसंशय यथार्थ आहे, अपूर्व असा आहे.

वास्तिवक पाहता विश्वात कुठेही अस्तित्वात नसलेली विषमता व मानवतेला कलंकीत करणारी धर्मव्यवस्था व वर्णव्यवस्था फक्त याच देशात आहे. जात, धर्म, पंध व वंशाच्या नावाखाली असमानता, अमानवता, अस्पृश्यता प्रस्तापित करून हजारो वर्षे या देशातील दलीत, शोषीत, वंचीत, आदिवासी, परीघावरील अन्य उपेक्षीत समाजाचा शोषन चालविला, त्यांच्यावर अन्याय, अत्याचार केले, पिडवणुक केली, त्यांचा अनिवत छड केला. या व्यवस्थेच्या विरूध्य अण्णाभाऊची लेखनी बंड पुकारते. साहित्याचा माध्यमातुन आंबेडकरी विचारधारेवर आधारीत समताधिष्ठीत नव समाजरचनेचे स्वप्न प्रत्येकाच्या मना—मनात परणा—या अण्णा भाऊचा जन्म दिनांक १ ऑगष्ट १९२० मध्येवाटेगाव येथील दिलत समाजात झाला. हा समाज आपलं जीवन गाव कुसावाहेर व्यथीत करीत होता. सत्ता, संपती व शिक्षणापासून वंचीत होता.

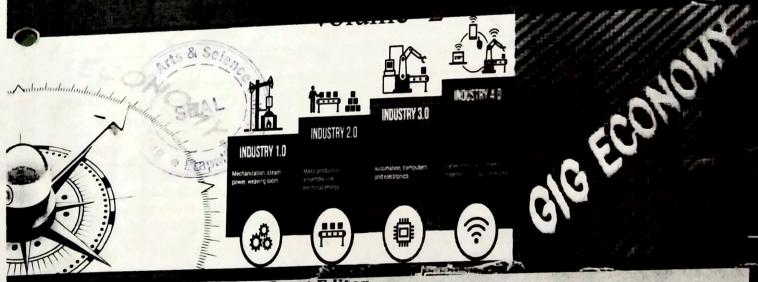
शिक्षणाशिवाय मनुष्य गुलाम बनतो. दुःख, दैन्य, दास्यातून संपूर्ण मनुष्यत्व येण्यासाठी शिक्षणाची नितात गरज आहे. याची जाणीव अण्णा भाऊंच्या विडलांना होती. म्हणून आपला मुलगा शिक्षण घेतला पाहीजे, प्रस्थापित व्यवस्थेशी बंड करण्यासाठी म्हणून अण्णा भाऊंना शाळेत घातले. परंतू वर्ण व्यवस्था त्यांच्याशिक्षणाच्या आड आली.

''गुरूवीना मार्ग नाही'' यासुभाषीतेला कालबाहय ठरवत स्वकर्तृत्वाच्या बळावर एकलव्यानी धनुर्विद्या शिकलेली असतांना अंगूठा मागून त्यांची विद्या हिस्कावारि द्रोणाचार्य, शुद्रांना विद्यार्जनाचा अधिकार नसल्याच्या सबबीखाली शंबुकाचे मुंडके उडविणारे प्रभू श्रीराम,

Etapali



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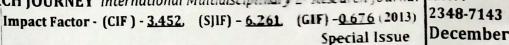


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डॉ.बाबासाहेब आंबेडकरांचे अर्थशास्त्रीयदृष्टिकोण प्रा. निलेश अरूण दूर्गे

अर्थशास्त्र विभाग भगवंतराव कला व विज्ञान महाविद्यालय एटापल्ली, जि.गडचिरोली

भारतरत्न डॉ.बाबासाहेब आंबेडकरांना आपण घटनातज्ञ, कायटयांचे गांढे अध्यासक, संसदपटू राजकारणी अशा विविध रुपात ओळखतो. पण एक अर्थतज्ञ म्हणून त्यांची ओळख फार दूर्मिळ अशी आहे. डॉ.बाबासाहेब आंबेडकरांच्या समूध्द आणि विविधांगी व्यक्तिमत्वाचा एक पैलू म्हणजे ते जागतिक किर्तीचे व्यासंगी अर्थतज्ञ होते. डॉ.बाबासाहेब आंबेडकरांनी स्वतःला मानवमूक्तीच्या लढयात झोकुन दिल्यामुळे त्यांचे सामाजिक, राजिकय, आणि धार्मिक विचार जितके समाजाला समजते त्या प्रमाणे आर्थिक विचार समजले नाहीत. डॉ.बाबासाहेब आंबेडकरांच्या विचारांचा त्या त्या कालावधीत विचार होऊन त्या आधारावर आर्थिक नियोजनाची अमंलबजावणी झाली असती तर आज भारताचा वेगळा नकाशा जगाला पहायला मिळाला असता.

डॉ.बाबासाहेब आंबेडकरांचे आर्थिक विचार

१) आर्थिक नियोजन

डॉ.बाबासाहेब आंबेडकरांच्या मते, आर्थिक विकासात उद्दीष्ट साध्य करण्यासाठी देशाने एक उचित आर्थिक नियोजनाची पघ्दत स्विकारणे आवश्यक होते. जवळपास १९० वर्षाच्या ब्रिटीश राजवटीतील पिळवणूकीनंतर आणि व्वितीय महायुघ्दातील भयंकर संहारानंतर स्वतंत्र झालेल्या भारताने स्वातंत्र्य लढयात अग्रेसर राहिलेल्या भारतीय राष्ट्रीय कांग्रेस पक्षाच्या आणि प्रथम पंतप्रधान जवाहरलाल नेहरुंच्या प्रेरणादायी नेतृत्वात आर्थिक नियोजनाशी आपली प्रगत व्हायला पाहिजे होती, पण ती आर्थिक नियोजन नसल्याने झालेली नाही. पण डॉ.बाबासाहेब आंबेडकर स्वतंत्र भारताचे पहिले मजुरमंत्री या नात्याने त्यांनी स्वतःच्या देशाच्या विकासाकरिता आर्थिक नियोजनाची काही वैशिष्टये ठळकपणे मांडली, त्यामध्ये भारतातील वाढत्या लोकसंख्येंच्या गरजा भागवण्यासाठी अन्नधान्यचे उत्पादन वाढव करणे, राष्ट्रीय उत्पन्नात भर घालणे, भविष्यातील आर्थिक विकासाचा पाया असणान्या औदयोगीकरणाचा वेग वाढवणे, श्रीमक आणि सुशिक्षीत तरुणांसाठी अनेकाअनेक रोजगाराच्या संधी उपलब्ध करुण देणे, तंत्रज्ञानाचा वापर करुन अधिकाअधिक रोजगार निर्माण करणे, ग्रामीण भागातील लघू व कुटीर उदयोगांना विशेष सवलती देवून आर्थिक सन्ता व संपन्तीचे विकेदीकरण करणे, आणि शहरी व ग्रामिण भागातील आर्थिक असमानता दुर करुण सामाजिकदुष्टया मानासलेल्या दलित, आदिवासी, ओ.बी.सी., अल्पसंख्याक व महिला वर्गाच सर्वांगीण विकासाकरीता संधी उपलब्ध करुन देणे आणि स्वावलबन व गरीबी दुर करणे इत्यादीचा समावेश आर्थिक नियोजनात सुचविले.

२) औद्योगिक पायाभरणी

डॉ.बाबासाहेब आंबेडकर बॉम्बे लेजिस्लेटिव्ह असेम्ब्लींचे सदस्य असताना (१९२६) प्रामीण भागातील गरिबांच्या समस्यांविषयीचे त्यांचे समग्र आकलन त्यांनी उभारलेल्या जनआंदोलनांमध्ये प्रतिबिंबित होते. शेतीमधील खोती पध्दतीविरूध्द त्यांनी केलेल्या यशस्वी आंदोलनामुळे अनेक प्रामीण गरीबांची आर्थिक शोषणातून मुक्तता झाली. महार पतन या नावाखाली सुरू असलेल्या शुध्द गुलामगिरीविरूध्द त्यांनी आवाज उठिविल्यानंतर प्रामीण भागातील गरिबांचा मोठा वर्ण शोषणमुक्त झाला. सावकारांच्या मनमानीला चाप लावण्यासाठी त्यांनी असेम्ब्लीमध्ये विधेयक आणले. औद्योगिक कामगारांच्या क्षेत्रात डॉ.आंबेडकरांनी १९३६मध्ये स्वतंत्र मजूर पक्षाची स्थापना केली. त्याकाळी कामगारांचा आवाज बुलंद करणाऱ्या अन्य संघटना होत्याच, मात्र त्यांना अस्पृश्य कामगारांच्या मानवाधिकारांशी काहीही देणे घेणे नव्हते. नव्या राजकीय पक्षाने ही उणीव भरून काढली. त्याचप्रमाणे व्हॉडमराँयज एक्झिक्युटिव्ह कौन्सिलचे कामगार सदस्य या नात्याने १९४२ ते १९४६ या काळात डॉ.बाबासाहेब आंबेडकर यांनी कामगारविषयक धोरणात आमूलाग्र सुधारणा घडवून आणल्या. त्यांत सेवायोजन कार्यालयाची स्थापना ही महत्वपूर्ण घटना होती आणि स्वतंत्र भारतातील औद्योगिक संबंधांची तीच पायाभरणी ठरली. डॉ.बाबासाहेब आंबेडकरांनी पाटबंबारे, ऊर्जा आणि इतर सार्वजनिक बांधकामे ही खातीही सांभाळली. देशाचे पाटबंधारे धोरण निश्चत करण्यात त्यांनी महत्वाची भूमिका बजावली, त्यांमध्ये दामोदर व्हेली प्रकल्पाचा यात प्राधान्याने समावेश करावा लागेल.

३) कुषी विचार

डॉ. आंबेडकरांनी शेतीतील उत्पादन वाडंगाटो यत्रप्रधान शेती, विस्तूत शेती, सहकारी व सामूदायीक शेतीचा पूरस्कार केला. त्यांच्यामते जिमन मालकाकडून शेतीचा विद्यादीचे हक्क काढून घेवून त्या बदल्यात त्यांना नूकसान भरपाई दयावी. जिमनीचे एकत्रिकरण करून योग्य अच्छानच्या जनीनी तयार करून. सामूहीक शेती विकसीत करावीत. जात, धर्म